



WHY SHOULD WE BE CONCERNED ABOUT PFAS AND OTHER EMERGING CONTAMINANTS?

ROCKY MOUNTAIN EHS PEER GROUP

APRIL 18, 2019

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Emerging Contaminants

Emerging contaminants are chemicals that have been detected in global drinking water supplies at trace levels and for which the risk to human health is not yet known.

- Past emerging contaminants
 - PCBs
 - Asbestos
 - MTBE
- “Emerging” emerging contaminants
 - Pharmaceuticals, personal care products, pesticides, herbicides and endocrine disrupting compounds
 - 1,4 dioxane
 - PFAS



PFAS: Per- (and Poly-) Fluoro Alkyl Substances
PFCs: Per- (and Poly-) Fluorinated Compounds
(also PerFluoroCarbons)

Why Do We Care About PFAS?

THE DENVER POST

NEWS > HEALTH

Toxic chemicals tainting Colorado groundwater also found in fast-food packaging

Extent of leaching from packaging to food not established

By BRUCE FINLEY | bfinley@denverpost.com | The Denver Post
PUBLISHED: February 1, 2017 at 12:01 am | UPDATED: February 1, 2017 at 3:02 pm

EPA United States Environmental Protection Agency Office of Water Mail Code 4304T EPA 822-R-16-005 May 2016

Drinking Water Health Advisory for Perfluorooctanoic Acid (PFOA)

Drinking Water Health Advisory for Perfluorooctanoic Acid (PFOA) – May 2016

MPRnews Sections Members More

Live updates: April snow storm continues across Minnesota and western Wisconsin

Minnesota settles water pollution suit against 3M for \$850 million

The Intercept

WITH NEW EPA ADVISORY, DOZENS OF COMMUNITIES SUDDENLY HAVE DANGEROUS DRINKING WATER

Sharon Letner
May 19 2016, 12:36 p.m.

SECTIONS New York Law Journal

The Emerging Crisis of PFAS Exposure

Paul J. Napoli and Tate J. Kunkle, New York Law Journal
October 6, 2017 | 0 Comments

THE DENVER POST Today Subscribe Newsletter

NEWS > ENVIRONMENT

Drinking water in three Colorado cities contaminated with toxic chemicals above EPA limits

80,000 people south of Colorado Springs being warned of high levels of PFCs in water

NEWS > ENVIRONMENT

North metro Denver groundwater contaminated with PFCs is flowing into a drinking-water system that supplies 50,000 residents

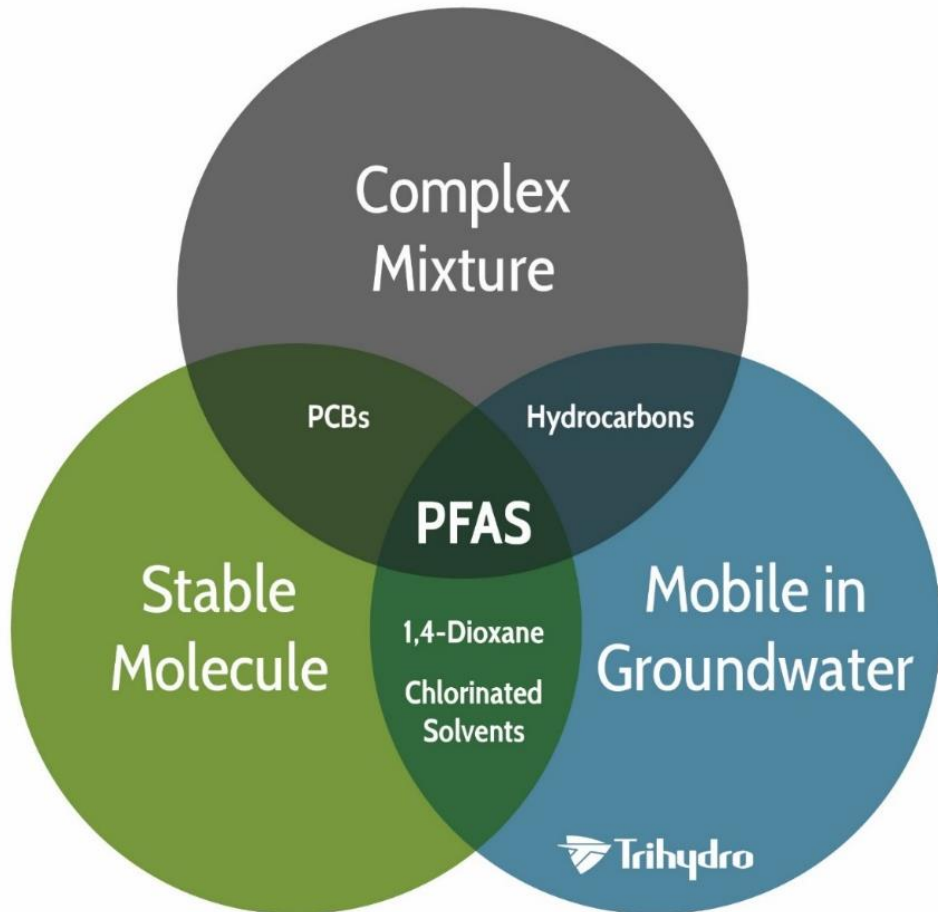
South Adams County Water and Sanitation District tests revealed elevated PFCs

HOME SEARCH The New York Times

U.S.

Tainted Water Near Colorado Bases Hints at Wider Safety Concerns

Why Do We Care About PFAS?



- **Site Characterization**
 - Presence in common sampling materials (Teflon...)
 - Low levels of interest (70 ppt)
 - Complex mixture (3000+ compounds)
- **Remediation**
 - Transformable, not degradable
 - Complex sorption (polar/non-polar)
- **Fate and Transport**
 - High solubility
 - Non-volatile
 - Atmospheric transport on regional/global scales
 - Persistence

Perfluorinated chemical uses

- PFAS Containing Materials
- Consumer products
 - Stain/water resistant fabrics
 - Personal care products
 - Grease resistant food packaging
- Industrial applications
 - Mist suppressants
 - Aerospace
 - Electronics
- Aqueous Film Forming Foam (AFFF)



PFAS: Key Terms

- **AFFF: Aqueous Film-Forming Foams**
- **Precursor:** A *polyfluorinated* compound that can biotransform into a *perfluorinated* compound
- **Short Chain vs Long Chain**
 - Long chain $\geq C6$ for PFSA's (e.g., PFOS),
 - $\geq C8$ for PFCAs (e.g., PFOA)



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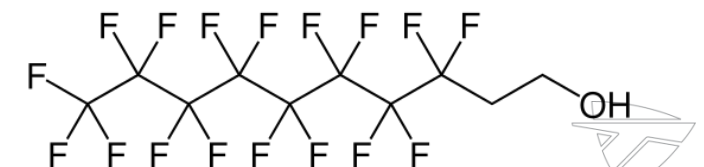
Perfluorooctane sulfonate (PFOS)



Perfluorooctanoic acid (PFOA)



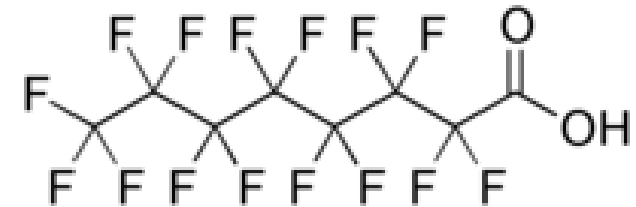
8:2-FTA (precursor)



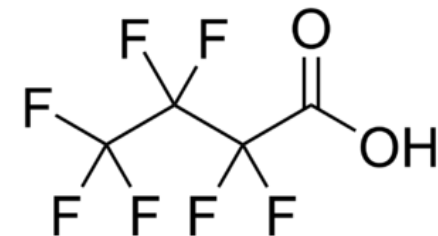
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Perfluorooctanoic acid (PFOA)



Perfluorobutanoic acid (PFBA)



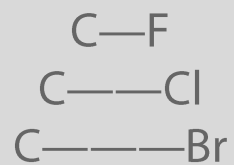
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PFAS



Why are PFAS so stable?



F-



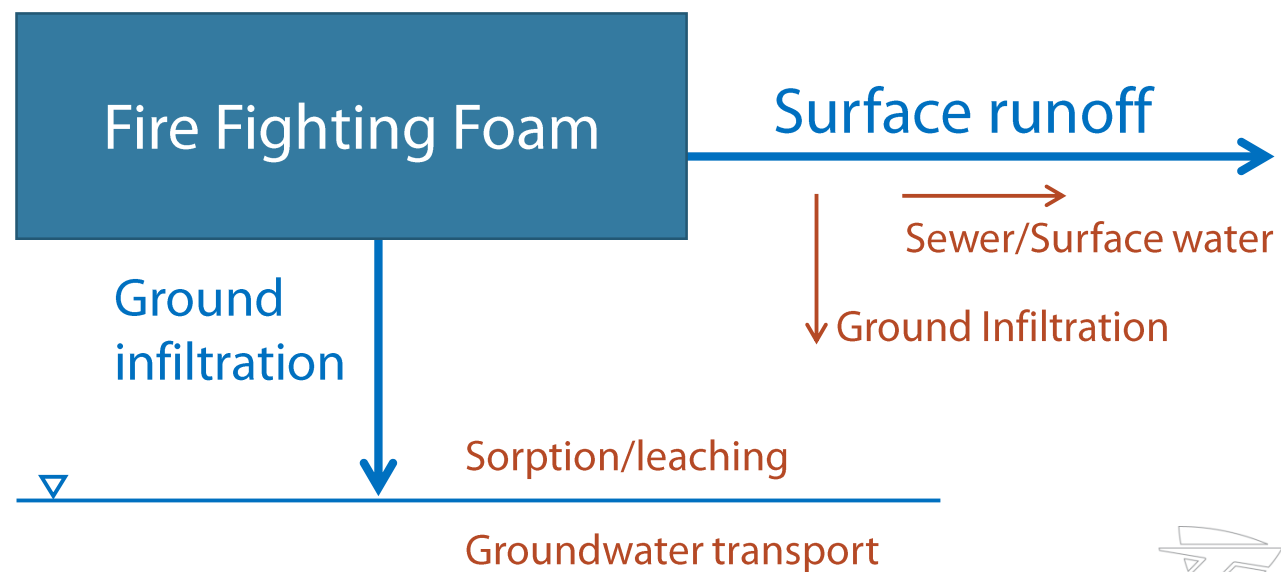
Cl-



Br-

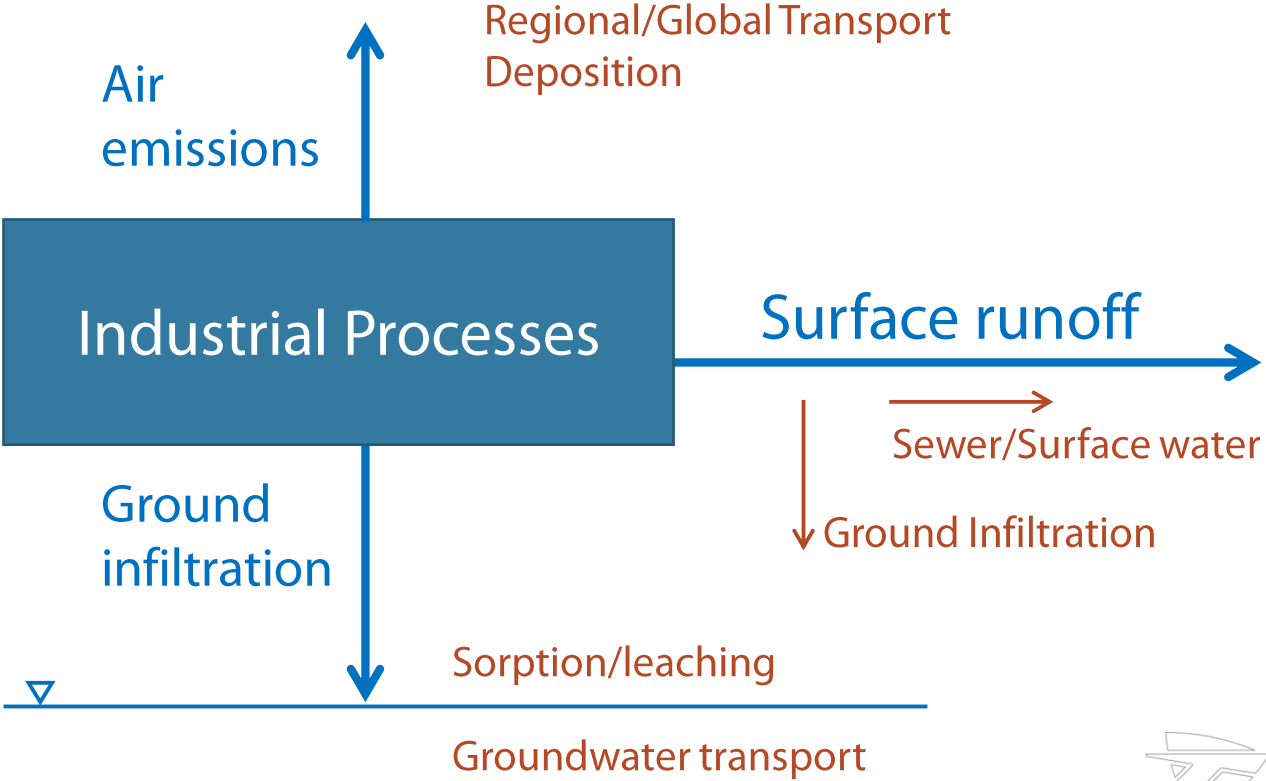
Environmental PFAS Sources

- Fire fighting
- Industrial Sources
- Landfills
- Water Treatment



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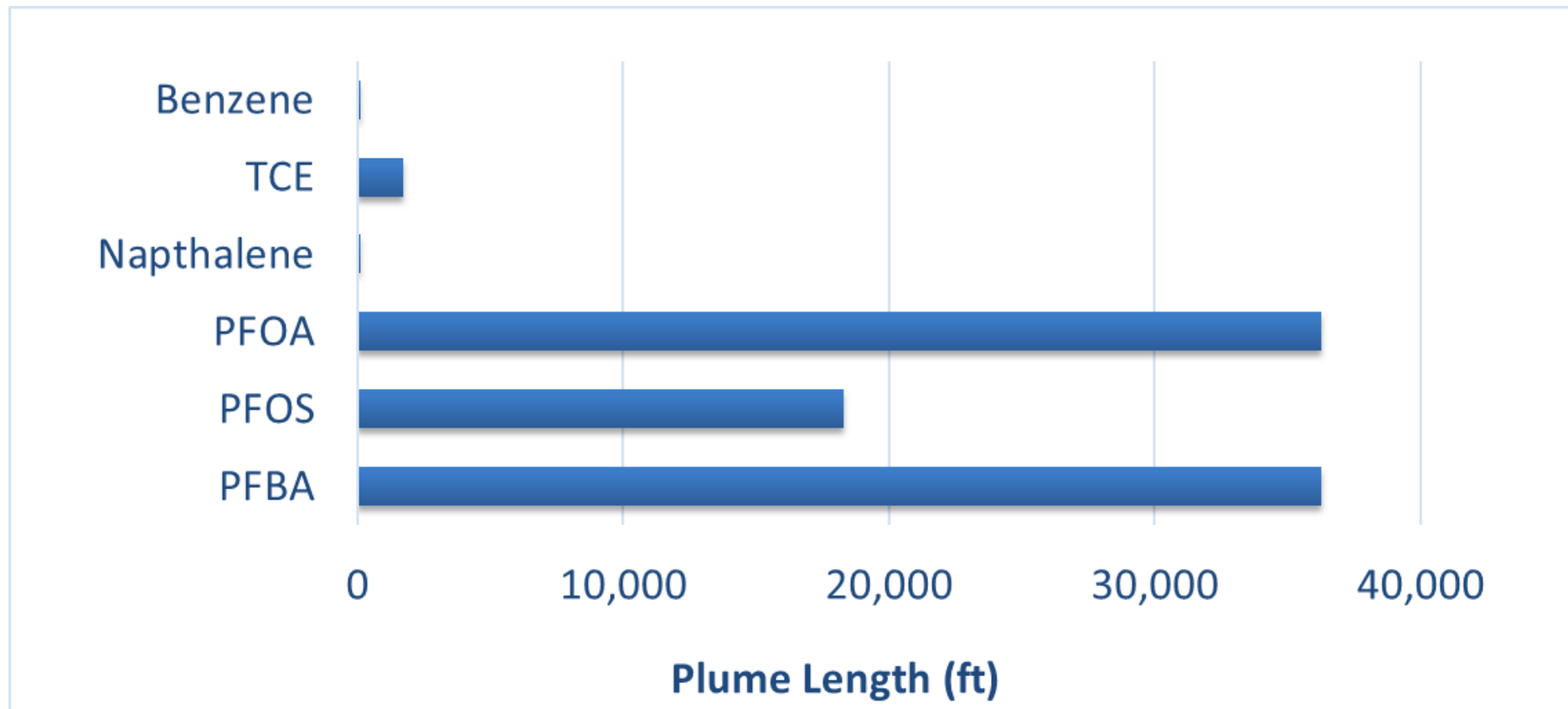


PFAS Fate and Transport

Key Considerations of PFAS in the Environment

- **Mobility**
 - **High solubility**
 - Sorption is complex (polar/non-polar)
 - PFAS compound of concern (PFOA, PFOS) are **non-volatile**
 - Atmospheric transport occurs on regional/global scales
- **Biotransformation**
 - Bioaccumulative
 - Precursor compounds may biotransform to PFOA, PFOS
 - **PFAS, as a whole, are not biodegradable**

Groundwater Plume Length



Site Characterization

PFAS Considerations

- PFAS-specific requirements for sampling and analysis (**no Teflon!**)
- QA/QC is very important for PFAS
- What PFAS to include in analysis?
 - Short list: 6 compounds on UCMR3 list
 - EPA 537: 14 compounds
 - Typical methods offer 25+ compounds
 - “Total PFAS” methods
 - TOP Assay
 - Total Organofluorine...Particle Induced Gamma Emissions (PIGE)



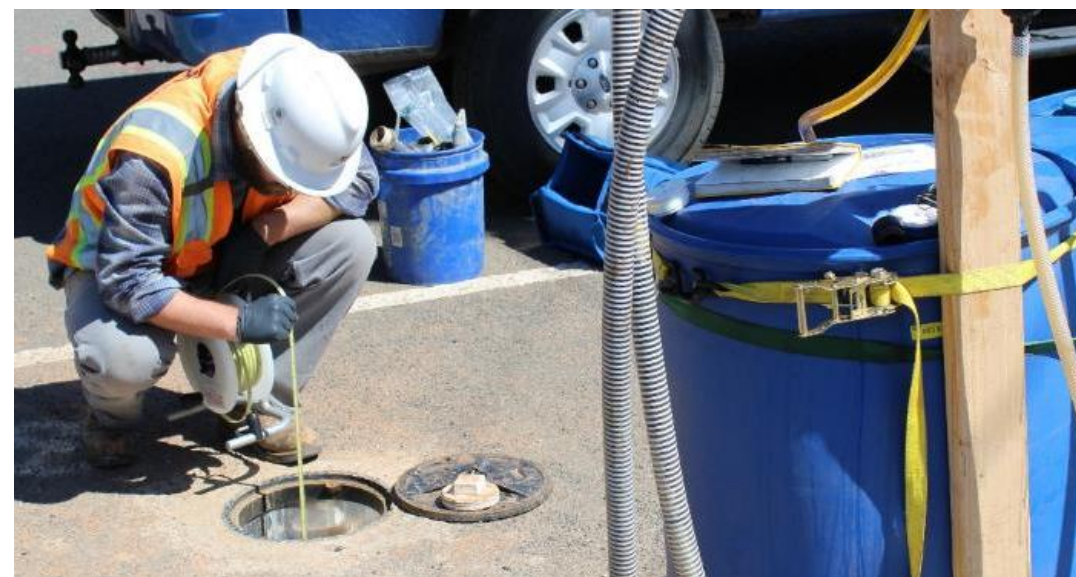
“Total” PFAS method: Total Oxidizable Precursor (TOP) assay

- Duplicate samples collected
- One is oxidized, converting many precursors to PFAAs on EPA 537 list
- Both are analyzed via EPA 537

Remediation

PFAS Considerations

- Considerations
 - Degradation challenge
 - Large number of individual compounds
- In situ PFAS degradation is (currently) not feasible
- Groundwater extraction/point of use treatment is typical (currently...)
 - GAC, ion exchange (consider long- vs. short-chain PFAS)
 - Reverse osmosis, nanofiltration

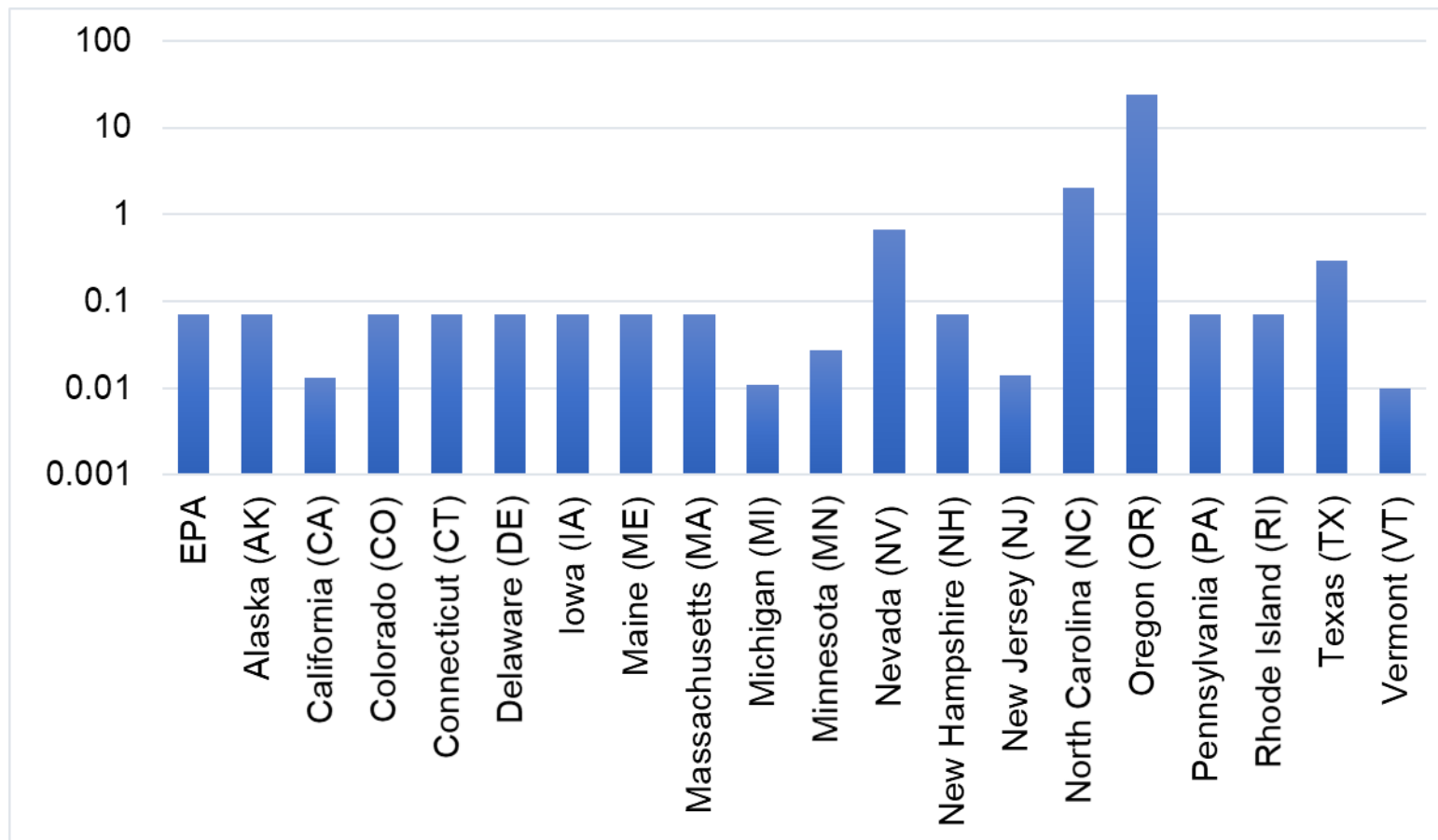


PFAS Remediation Considerations

- Resistant to air stripping & degradation
- Some PFAS partially degradable, PFAS as a whole are not
- Treatment of concentrated waste streams

PFAS: Regulatory Status

- EPA Health Advisory (HA) Levels
 - [PFOA] + [PFOS] = 70 parts per trillion (ppt)
- No EPA reg's on short-chain PFASs or precursors
- 19 state-specific MCLs or Groundwater Quality Standards
 - ITRC table (updated quarterly)



<https://pfas-1.itrcweb.org/fact-sheets/>

PFAS: Regulatory Status

Colorado Dept of Public Health and Environment (CDPHE)

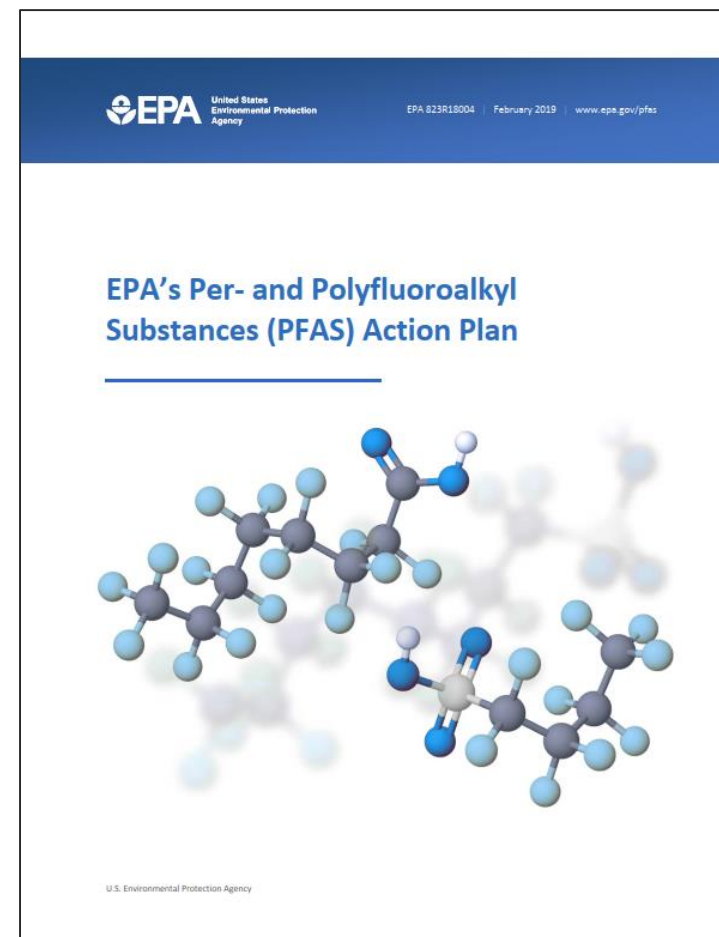
State of Colorado

- Has adopted the US EPA's (non enforceable) HA level of 70 ppt
- Groundwater Quality Standard of 70 ppt adopted for El Paso County only
- CDPHE has committed to conducting statewide assessment of potential PFAS sources (mentioned this in EPA public forum in Colorado Springs, 8/7-8/8)
 - "CDPHE's current priority is understanding the potential impacts on drinking water sources from PFC releases. Once we have a better understanding of the potential extent of any contamination, we will evaluate the need for any statewide groundwater standards."

EPA Activities

EPA PFAS Action Plan released February 2019

- Drinking water MCL (review 2019)
- ‘Hazardous substance’ designation
Under CERCLA (started 2018)
- Toxics release inventory (start 2019)
- Drinking water monitoring (2020)
Public water supply sampling under UCMR
- Research (ongoing)
Analysis, remediation, toxicity
- Enforcement (ongoing)
- Risk communication (2019)



PFAS at Air Force Facilities

203 Installations Require Preliminary Assessments (PAs)



*Estimated completion for last base, Joint Base Elmendorf-Richardson, Alaska, is Spring 2020.

of those 203, **189** Installations Expected to Require a Site Inspection (SIs)



IDENTIFY

- 39 BRAC
 - 89 Active/Reserve
 - 75 ANG
-
- 30 BRAC
 - 84 Active/Reserve
 - 75 ANG

RESPOND



Air Force Drinking Water Mitigation:

- Bottled water
- Point-of-use filtration
- Whole-house filtration
- Municipal water supply hookup

BRAC

- K.I. Sawyer
- March
- Mather
- Pease
- Plattsburgh
- Reese
- Wurtsmith

ACTIVE/RESERVE

- Cannon
- Dover
- Eielson
- Ellsworth
- Fairchild
- JB McGuire-Dix-Lakehurst
- Mountain Home
- New Boston
- Peterson
- Wright-Patterson

ANG

- JB Cape Cod
- Gabreski
- Horsham
- Toledo

PREVENT

176/176
Installations transitioned to new C6 AFFF in stockpiles and fire trucks

\$5.3M
Cost for ecologic system kits for fire vehicles

93%
Installations finished retrofitting vehicles with ecologic system kits

979,000
Gallons of legacy AFFF incinerated

\$10.8M
Cost to date to replace and incinerate legacy AFFF in stockpiles and fire trucks



\$321M* total expenditure

PFOS/PFOA Actions to Date investigations | mitigations | prevention
*through end of 1st Qtr FY19



The Future of AFFF



“The environmentally-mindful CHEMGUARD C606-MS Concentrate formulation contains **short-chain, C-6 fluorochemicals** manufactured using a telomer-based process. The telomer process **produces no PFOS, and these C-6 materials do not breakdown to yield PFOA**. The fluorochemicals used in the concentrate meet the goals of the U.S. Environmental Protection Agency 2010/15 PFOA Stewardship Program.”

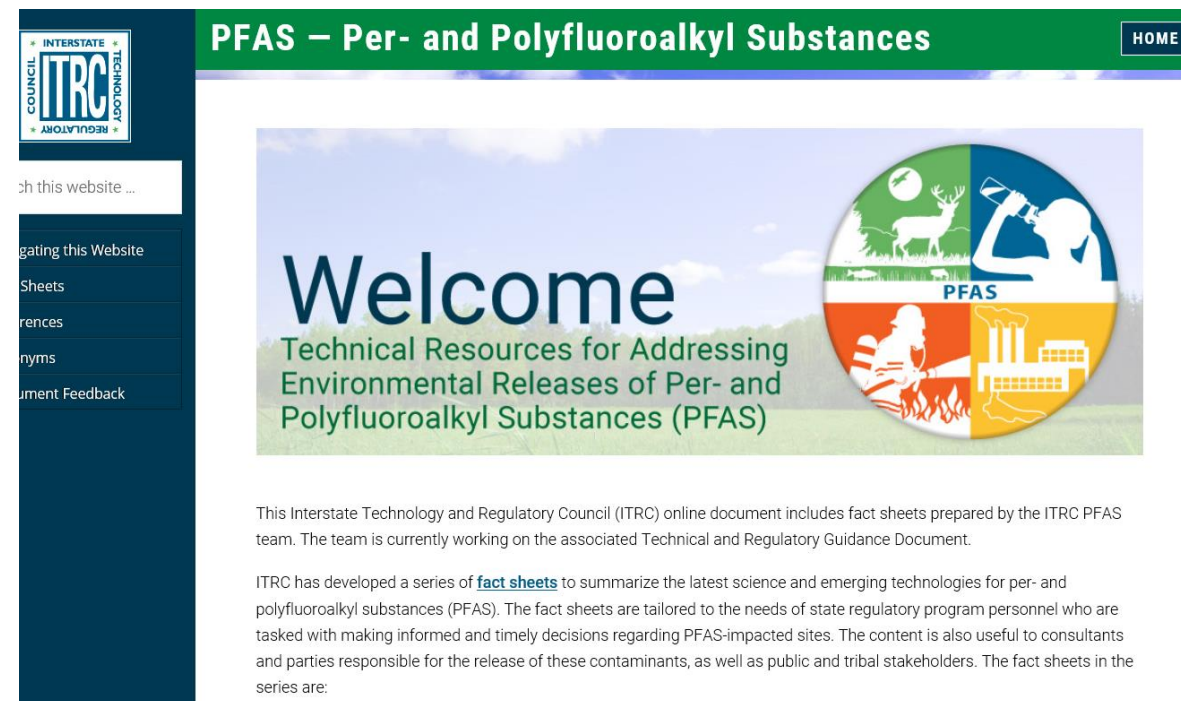


“Claims by industry that the shorter chain-length compounds are less toxic should not be accepted just because it is a convenient way to allow them to **easily replace the problematic products**. After all, the shorter-chain compounds are more mobile in groundwater and are nearly impossible to remove during drinking water treatment,” *Sedlak, 2016, Environmental Science & Technology, Fool me Once*

Guidance Documents

Interstate Technology Regulatory Council (ITRC)

1. Naming Conventions and Physical and Chemical Properties
2. Regulations, Guidance, and Advisories
3. History and Use
4. Environmental Fate and Transport
5. Site Characterization Considerations, Sampling Precautions, and Laboratory Analytical Methods
6. Remediation Technologies and Methods
7. Aqueous Film Forming Foam



Interstate Technology Regulatory Council (ITRC) logo

PFAS – Per- and Polyfluoroalkyl Substances

Welcome

Technical Resources for Addressing Environmental Releases of Per- and Polyfluoroalkyl Substances (PFAS)

This Interstate Technology and Regulatory Council (ITRC) online document includes fact sheets prepared by the ITRC PFAS team. The team is currently working on the associated Technical and Regulatory Guidance Document.

ITRC has developed a series of **fact sheets** to summarize the latest science and emerging technologies for per- and polyfluoroalkyl substances (PFAS). The fact sheets are tailored to the needs of state regulatory program personnel who are tasked with making informed and timely decisions regarding PFAS-impacted sites. The content is also useful to consultants and parties responsible for the release of these contaminants, as well as public and tribal stakeholders. The fact sheets in the series are:

<https://pfas-1.itrcweb.org/fact-sheets/>

Why do we care about PFAS?

- PFAS compounds are “ideal” contaminants
 - The issue is bigger than just PFOA and PFOS
 - Complex mixture and unique chemistry
 - Extremely low groundwater concentrations of interest (ppt)
- Ongoing challenges
 - Regulatory uncertainty
 - Uncertain/developing understanding of toxicology
 - Analysis is highly specialized
 - Remediation options exist, but are currently limited

Emerging Contaminants – 1,4-Dioxane

- **Presence in Environment**
 - Found in groundwater sites throughout US
 - Used in paint strippers, dyes, greases, varnishes and waxes
 - Manufacture of polyethylene terephthalate (PET) plastic
 - Purifying agent in pharmaceuticals
 - By-product in personal care products, shampoos, cosmetics
 - **Mostly associated with chlorinated solvents**
 - 90% of production in 1970s
 - Stabilizer for 111-TCA



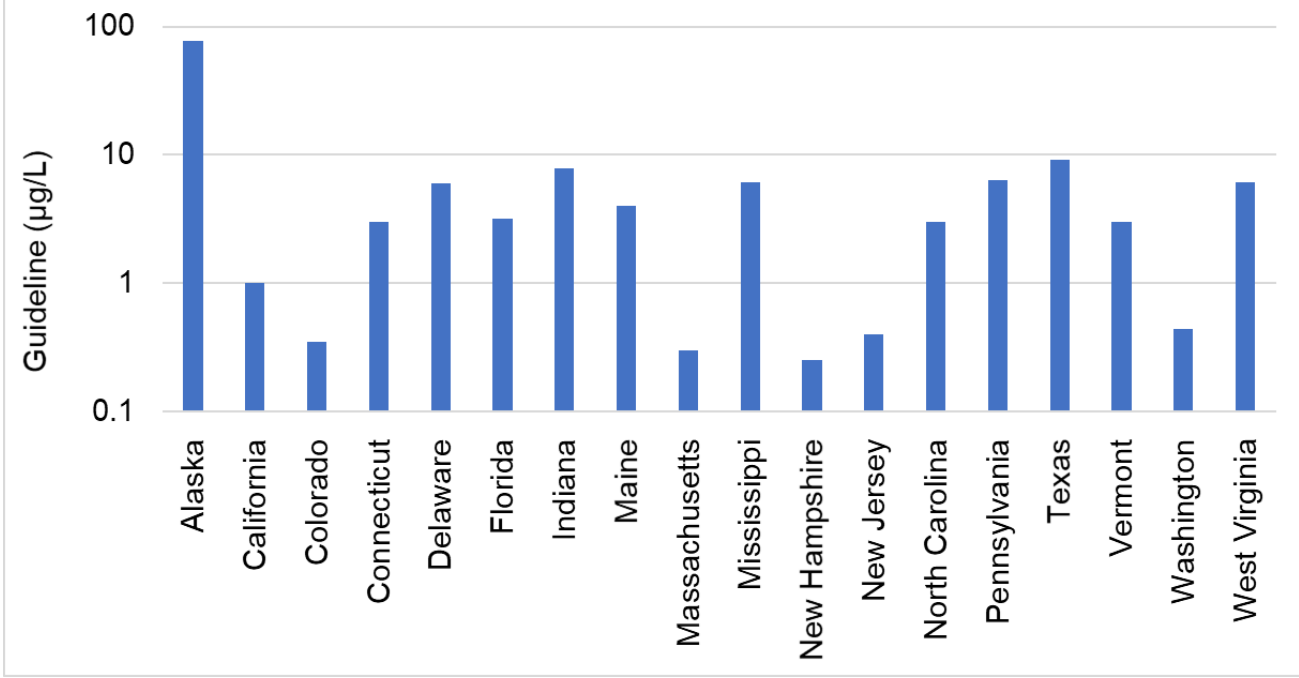
Toxicology Considerations

- Exposure
 - Ingestion of contaminated food and water, or dermal contact
 - Worker exposures via inhalation of vapors
- Health impacts
 - Classified by EPA as “likely to be carcinogenic to humans”
 - Short-term exposure may cause eye, nose and throat irritation
 - Exposure to high may result in nausea, drowsiness, headache, and irritation of the eyes, nose and throat
 - Long-term exposure may cause kidney and liver damage
- Does not bioaccumulate, biomagnify, or bioconcentrate in the food chain

For more information refer to EPA document: Technical Fact Sheet – 1,4-Dioxane. November 2017

1,4-Dioxane Regulatory Standards for Groundwater

- No Federal Maximum Contaminant Level; Health Advisory - 0.35 µg/L
- USEPA Regional Screening Level – 0.46 µg/L
- State Standards – First proposed by Colorado in 2005 at 6.1 µg/L; since lowered to 0.35 µg/L (CDPHE 2012)



Variability > 2 orders of magnitude
Range: 0.25 µg/L (New Hampshire) to 77 µg/L (Alaska)



Remediation Options

Ex-Situ

- Air stripping
- Granular activated carbon (GAC)
- Synthetic resins
- Advanced oxidation
- Aerobic bioreactor
- Other (e.g., electrolytic)

In-Situ

- Physical
 - Air sparging / SVE
 - Thermal ?
- Chemical
 - ISCO
 - Chemical reduction (ISCR)
- Biological
 - Aerobic biostimulation
 - Anaerobic biostimulation
 - Co-metabolic processes
- Monitored Natural Attenuation



So...why do we care about 1,4-dioxane?

- Widespread occurrence in drinking water
- Very high affinity for water: high mobility, difficult removal
- Remediation challenges: is degradable (or extractable) under certain conditions
- Regulatory uncertainty: typical for emerging contaminants
- Toxicology unknowns

- What's in store...
 - ITRC team developing for 2019-2020
 - ESTCP/SERDP funded project (Extreme SVE, mixed contaminants, electrolytic)

